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All matter consists of tiny particles that are in constant motion.

molecules/atom. the particles in a gas usually consist of. insignificant.

The particles in a gas are considered to be small, hard spheres with an _____ volume.

empty space ...

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cation. any metal in

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any atom or group of

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Atmospheric Pressure.

Barometer. Two or

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because of its motion.

All matter consists of
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to be small, hard sphereⁿ
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more different

molecular forms of the

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result from the collisions of atoms and molecules in air with a device that is used to measure atmospheric pressure. Allotrope. Two or more ...

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Energy. The energy an object has because of its motion. Kinetic Theory .

The theory that states that all matter consists of tiny particles that are in constant motion.

When applied to gases makes the following fundamental

assumptions: The particles in a gas are small hard spheres with insignificant volume,

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The motion of gas
particles is ...

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Binding, Vision and

Light Induced

Molecular Changes ...

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friends love reading?
Are you the one that
don't have such hobby?
So, it's important for
you to begin having that
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We're determined that
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